organic papers

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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.005 Å R factor = 0.057 wR factor = 0.181 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1-[(2,6-Dimethylphenyl)aminocarbonylmethyl]-4-{[3-(3-methoxyphenyl)-1,2,4-oxadiazol-5-yl]methyl}piperazine

The title compound, $C_{24}H_{29}N_5O_3$, contains intramolecular N-H···N hydrogen-bond and intermolecular C-H··· π interactions.

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Comment

Piperazine derivatives are of interest because of their biological properties. Some derivatives of piperazine have antifilarial, anti-amebic and spermicidal properties (Sonurlikar *et al.*, 1977). Some show high efficacy in treating or preventing neuronal damage or stimulating nerve growth (Tomlinson *et al.*, 2004). Some have also been used to treat psychosis and bipolar disorders (Aicher *et al.*, 2004) or act as neurokinin antagonists (Janssens *et al.*, 2004).



The molecular structure of (I) is shown in Fig. 1, where the dashed line indicates an intramolecular $N-H\cdots N$ hydrogen bond (Table 1). The bond lengths and angles are given in Table 1. In the crystal structure, there are also $C-H\cdots \pi$ interactions (Table 1 and Fig. 2).

Experimental

4-[(2,6-Dimethylphenyl)aminocarbonylmethyl]piperazine (20 mmol) was dissolved in acetone (20 ml) and potassium carbonate (30 mmol) was added. 3-(3-Methoxylphenyl)-5-chloromethyl-1,2,4-oxadiazole (20 mmol) in acetone (20 ml) was added to this mixture. The resulting mixture was refluxed for 6 h. Concentration of the mixture under reduced pressure afforded crude compound (I). Pure compound (I) was obtained by recrystallization from ethyl acetate. Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an ethanol solution. ¹H NMR (CDCl₃, p.p.m.): δ 8.56 (*m*, 1H), 7.68–7.70 (*m*, 1H), 7.61–7.62 (*m*, 1H), 7.37–7.40 (*m*, 1H), 7.03–7.10 (*m*, 4H), 3.96 (*m*, 2H), 3.86 (*s*, 3H), 3.22 (*m*, 2H), 2.78 (*m*, 8H), 2.21 (*s*, 6H).

Crystal data	
$C_{24}H_{29}N_5O_3$	$D_x = 1.262 \text{ Mg m}^{-3}$
$M_r = 435.52$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 25
a = 17.120 (3) Å	reflections
b = 10.271 (2) Å	$\theta = 9-12^{\circ}$
c = 13.042 (3) Å	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 91.10 \ (3)^{\circ}$	T = 293 (2) K
V = 2292.9 (8) Å ³	Block, colourless
Z = 4	$0.40 \times 0.40 \times 0.10 \text{ mm}$

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Data collection

Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{\min} = 0.967, T_{\max} = 0.992$ 4695 measured reflections 4484 independent reflections 2051 reflections with $I > 2\sigma(I)$

Refinement

Table 1

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.057$ $wR(F^2) = 0.181$ S = 0.954484 reflections 295 parameters H atoms treated by a mixture of independent and constrained refinement

Hydrogen-bond geometry (Å, °).

 $\begin{aligned} R_{\text{int}} &= 0.032\\ \theta_{\text{max}} &= 26.0^{\circ}\\ h &= -20 \rightarrow 20\\ k &= -12 \rightarrow 0\\ l &= 0 \rightarrow 15\\ 3 \text{ standard reflections}\\ \text{every 200 reflections}\\ \text{intensity decay: none} \end{aligned}$

$$\begin{split} &w = 1/[\sigma^2(F_{\rm o}^2) + (0.08P)^2 \\ &+ 0.15P] \\ &where \ P = (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ (\Delta/\sigma)_{\rm max} = 0.001 \\ \Delta\rho_{\rm max} = 0.15 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.16 \ {\rm e} \ {\rm \AA}^{-3} \end{split}$$

D−H···A	<i>D</i> -H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$N5-H5A\cdots N4$ $C15-H15B\cdots Cg4^{i}$	0.84 (3)	2.27 (3)	2.734 (4)	115 (3)
	0.97	2.79	3.520 (4)	133

Symmetry code: (i) -x, $y - \frac{1}{2}$, $-z + \frac{1}{2}$. Cg4 is the centroid of atoms C17–C22.

Atom H5A, attached to atom N5, was located in a difference Fourier map and refined freely. Other H atoms were placed in calculated positions with C–H distances in the range 0.93–0.97 Å and refined using a riding model with $U_{\rm iso}(\rm H) = 1.2U_{eq}(\rm C)$ or $1.5U_{eq}(\rm C_{methyl})$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

References

- Aicher, T. D., Chen, Z., Le Huerou, Y., Martin, F. M., Pineiro-nunez, M. M., Rocco, V. P., Ruley, K. M., Schaus, J. M., Spinazze, P. G. & Tupper, D. E. (2004). Intl Patent No. WO2004014895.
- Enraf-Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf-Nonius, Delft, The Netherlands.

Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.



Figure 1

A view of the molecular structure of (I); the dashed line indicates an intramolecular $N-H \cdots N$ hydrogen bond. Displacement ellipsoids for non-H atoms are drawn at the 30% probability level.



Figure 2 The C-H··· π interaction in (I), shown as a dashed line.

- Janssens, F. E., Sommen, F. M., De Boeck, B. C. A., Leenaerts, J. E., Van Roosbroeck, Y. E. M. & Diels, G. S. M. (2004). Intl Patent No. WO2004033428.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351– 359.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Siemens (1996). SHELXTL. Version 5.06. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sonurlikar, U. A., Shanker, B., Kirke, P. A. & Bhide, M. B. (1977). Bull. Haffkine Inst. 5, 94–96.
- Tomlinson, R., Lauffer, D. & Mullican, M. (2004). US Patent No. US2004034019.